Reactions of Carboxylic Acids and Nitriles

- **20.33** How could you convert butanoic acid into the following compounds? Wilte each step showing the reagents needed.
 - (a) 1-Butanol
- (b) 1-Bromobutane
- (c) Pentanoic acid

- (d) 1-Butene
- (e) Octane
- 20.36 How would you prepare the following compounds from benzene? More than one step is required in each case.
 - (a) m-Chlorobenzoic acid
- (b) p-Bromobenzoic acid
- (c) Phenylacetic acid, C₆H₅CH₂CO₂H
- **20.37** Predict the product of the reaction of *p*-methylbenzoic acid with each of the following:
 - (a) LiAlH₄, then H₃O⁺
- (b) N-Bromosuccinimide in CCl₄
- (c) CH₃MgBr in ether, then H₃O⁺
- (d) KMnO₄, H₃O⁺
- 20.52 *p*-Aminobenzoic acid (PABA) is widely used as a sunscreen agent. Propose a synthesis of PABA starting from toluene.

Problem 21.5

Predict the products of the following nucleophilic acyl substitution reactions:

(b)
$$O$$
 NH_3 ?

$$\frac{\text{Na}^{+}\text{-OCH}_{3}}{\text{CH}_{3}\text{OH}} \rightarrow ?$$

21.39 Predict the product(s) of the following reactions:

(a)
$$CO_2CH_2CH_3$$

(c)
$$COCI$$
 CH_3NH_2

(d)
$$CO_2H$$
 H
 CH_3
 CH_3OH
 H_2SO_4
?

(e)
$$CH_3$$
 $\frac{1}{1}$ $H_2C = CHCHCH_2CO_2CH_3$ $\frac{1. \text{ LiAlH}_4}{2. \text{ H}_3O^+}$

(g)
$$CONH_2$$
 $1. LiAlH_4$ $2. H_2O$

(h)
$$CO_2H$$
 $SOCI_2$ 7