### Problem 16.1

Monobromination of toluene gives a mixture of three bromotoluene products. Drawqiii name them.

# Problem 16.3

How many products might be formed on chlorination of o-xylene (o-dimethylbenzene), w-xylene, and p-xylene?

### Problem 16.5

Which of the following alkyl halides would you expect to undergo Friedel-Crafts reaction with rearrangement and which without? Explain.

- (a) CH<sub>3</sub>CH<sub>2</sub>Cl
- (b) CH<sub>3</sub>CH<sub>2</sub>CH(Cl)CH<sub>3</sub>
- (c) CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CI

- (d) (CH<sub>3</sub>)<sub>3</sub>CCH<sub>2</sub>Cl
- (e) Chlorocyclohexane

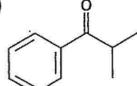
## Problem 16.6

What is the major monosubstitution product from the Friedel–Crafts reaction of benzene with 1-chloro-2-methylpropane in the presence of AICl<sub>3</sub>?

## Problem 16.7

Identify the carboxylic acid chloride that might be used in a Friedel–Crafts acylation reaction to prepare each of the following acylbenzenes:

(a)



(b)

# Problem 16.8

Rank the compounds in each of the following groups in order of their reactivity to electrons philic substitution:

- (a) Nitrobenzene, phenol, toluene, benzene
- (b) Phenol, benzene, chlorobenzene, benzoic acid
- (c) Benzene, bromobenzene, benzaldehyde, aniline

## Problem 16.9

Predict the major products of the following reactions:

- (a) Nitration of bromobenzene
- (b) Bromination of nitrobenzene
- (c) Chlorination of phenol
- (d) Bromination of aniline

# roblem 16.12

setamilide is less reactive than aniline toward electrophilic substitution. Explain.

# Broblem 16.14

A what position would you expect electrophilic substitution to occur in each of the followline substances?

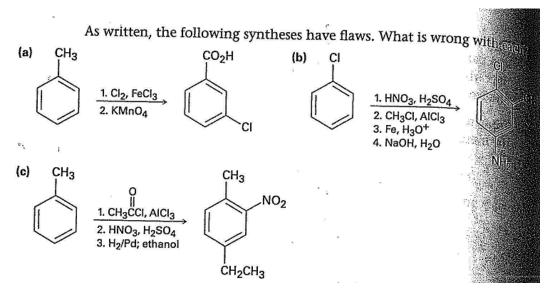
#### **Problem 16.39**

The carbocation electrophile in a Friedel–Crafts reaction can be generated a ways other than by reaction of an alkyl chloride with AlCl<sub>3</sub>. For example reaction of benzene with 2-methylpropene in the presence of  $H_3PO_4$  yields tert-butylbenzene. Propose a mechanism for this reaction.

## **Problem 16.43**

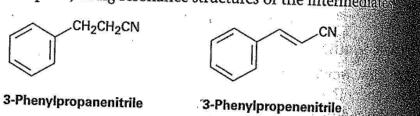
Using resonance structures of the intermediates, explain why bromination of biphenyl occurs at ortho and para positions rather than at meta.

#### **Problem 16.48**



### **Problem 16.50**

▲ Electrophilic substitution on 3-phenylpropanenitrile occurs at the original and para positions, but reaction with 3-phenylpropenenitrile occurs and meta position. Explain, using resonance structures of the intermediates



### **Problem 16.52**

At what position, and on what ring, would you expect the following substances to undergo electrophilic substitution?

**Problem 16.65** 

Propose a mechanism to account for the following reaction:

$$H_3C$$
 $CH_2CI$ 
 $AICI_3$ 
 $H_3C$ 
 $CH_3$ 
 $CH_3$ 
 $CH_3$